Ptable

“Explore Periodic table using Ptable”

Presented by

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❑ What is Ptable?

- **Ptable** is a website that offers a free, layered, and interactive periodic table.
- The main page displays the periodic table, which you can click to get Wikipedia definitions.
- You can learn about the 18 groups of elements, interactively explore 16 properties and their sub-properties, view images of orbitals, list selected or all isotopes, and iterate compounds, plus much more.
- It is an amazing educational website for young kids to learn the elements and older kids to help them with their chemistry classes.

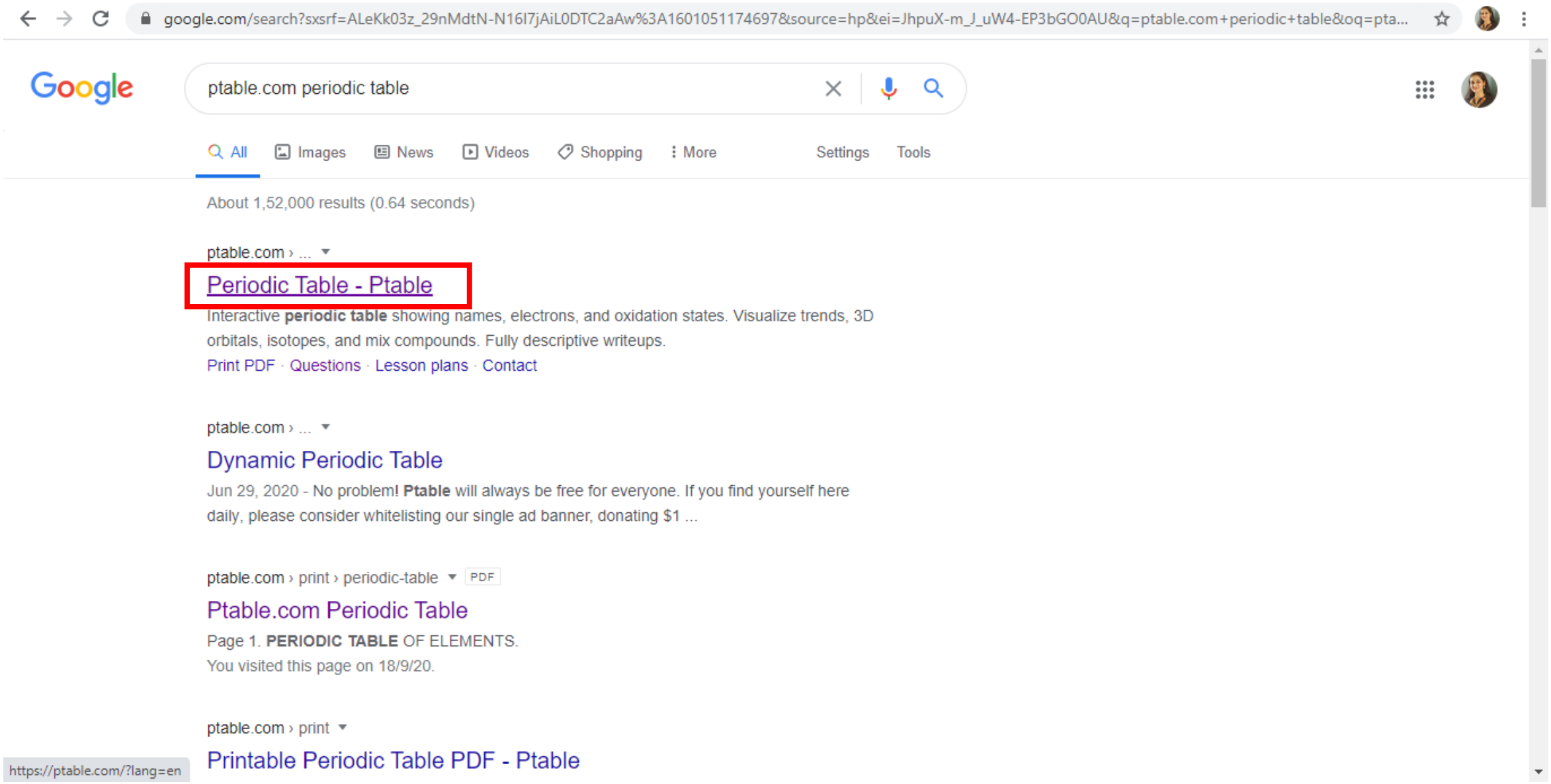
❑ Ptable in Education

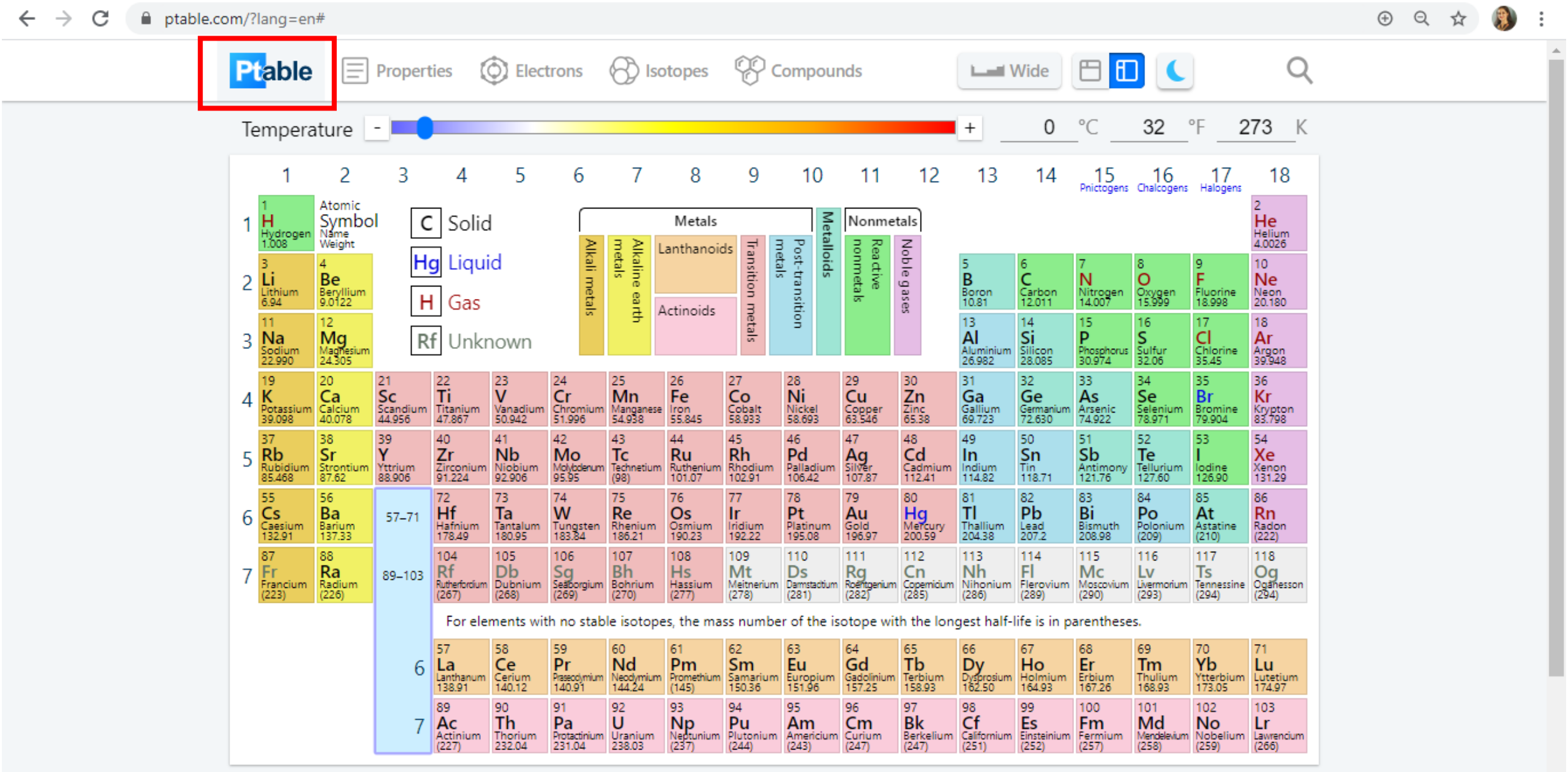
- Ptable is awesome for kids truly interested in chemistry or who have a parent who wants to get them interested.
- The sheer quantity of information about each element, from atomic number to abundance in the universe, is quite enough, but then add color codes, dynamic *and* interactive data, links to a complete set of videos and podcasts from distinguished sources, and striking and orderable images and posters, and you've got a website with some serious depth.

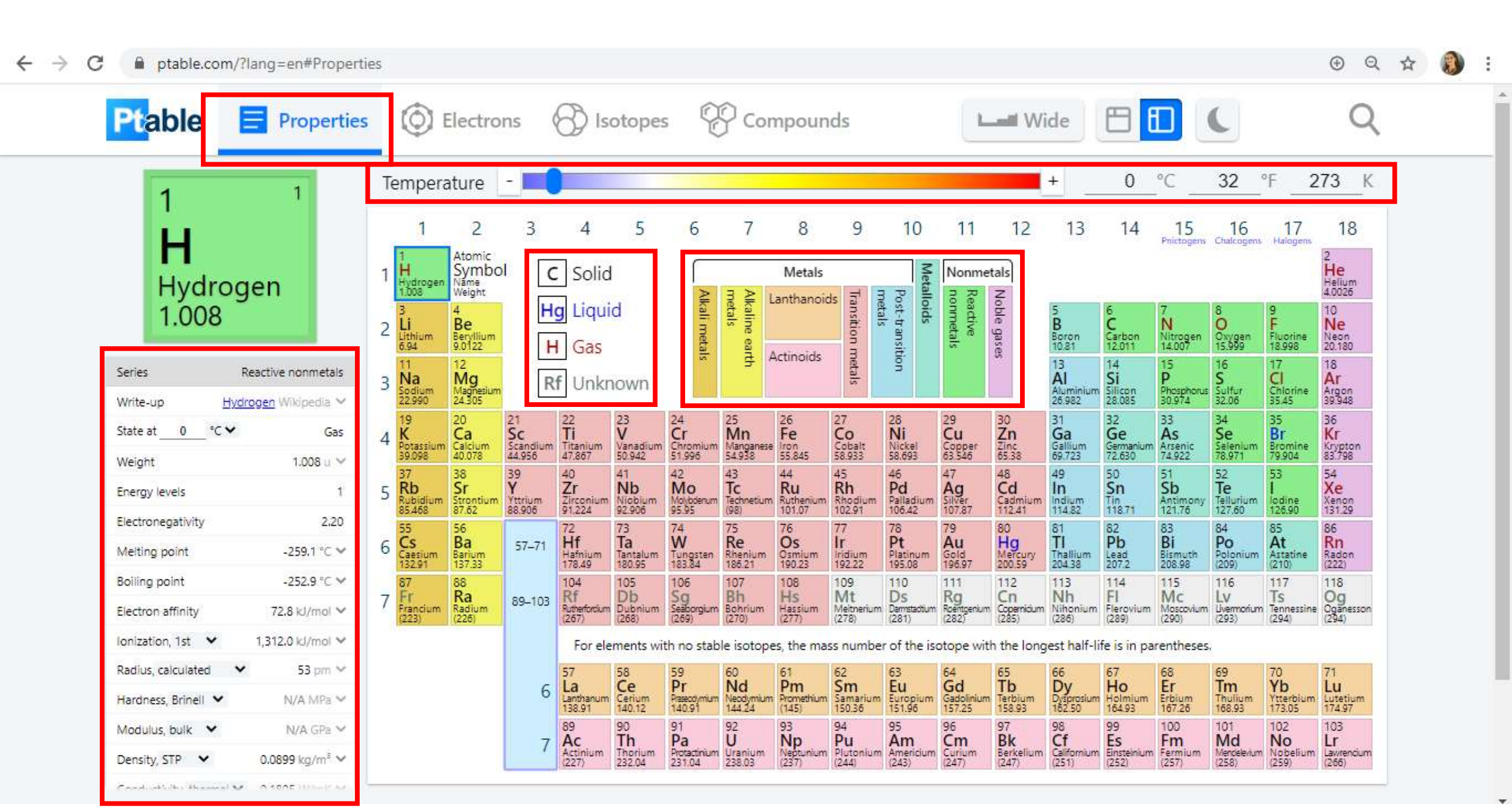
❑ Creator of Ptable

- **Michael Dayah** created the Ptable in October 1997, which holds copyright and trademark for its creation but is free to all for the use and application.
- Ptable, in Dayah's own words, makes "Mendeleev's creation come alive.“
- One can also install for computer or mobile and work offline. It also supports iOS and Android based devices and is quite adaptable.

❑ How to get started with Ptable on Computer?




ptable.com/?lang=en#
Ptable
Properties
Electrons
Isotopes
Compounds
Wide
Temperature
0 °C 32 °F 273 K
Atomic
Symbol
Name
Weight
C Solid
Hg Liquid
H Gas
Rf Unknown
Metals
Nonmetals
Alkali metals
Alkaline earth metals
Lanthanoids
Actinoids
Transition metals
Post-transition metals
Metalloids
Reactive nonmetals
Noble gases
Pnictogens
Chalcogens
Halogens
For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.



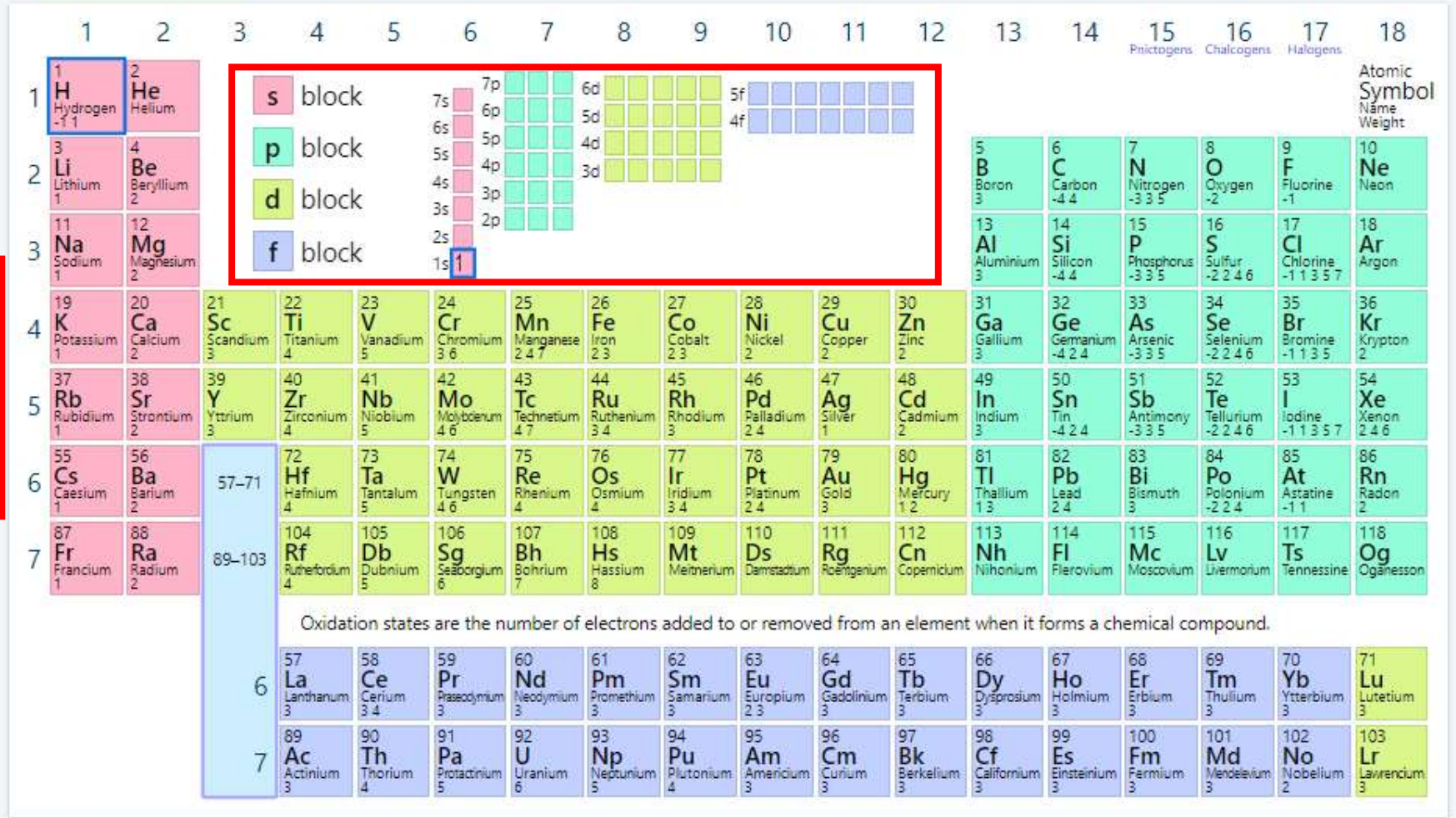
PROPERTIES - It displays the physical and chemical properties of the elements

ptable.com/?lang=en#Electrons

Ptable | Properties | Electrons | Isotopes | Compounds | Wide

1 1
H
Hydrogen
1.008

Oxidation states	-1, 1
Configuration	$1s^1$
Expanded	$1s^1$
Energy levels	1
Quantum numbers	$l=0, m=0, n=1$



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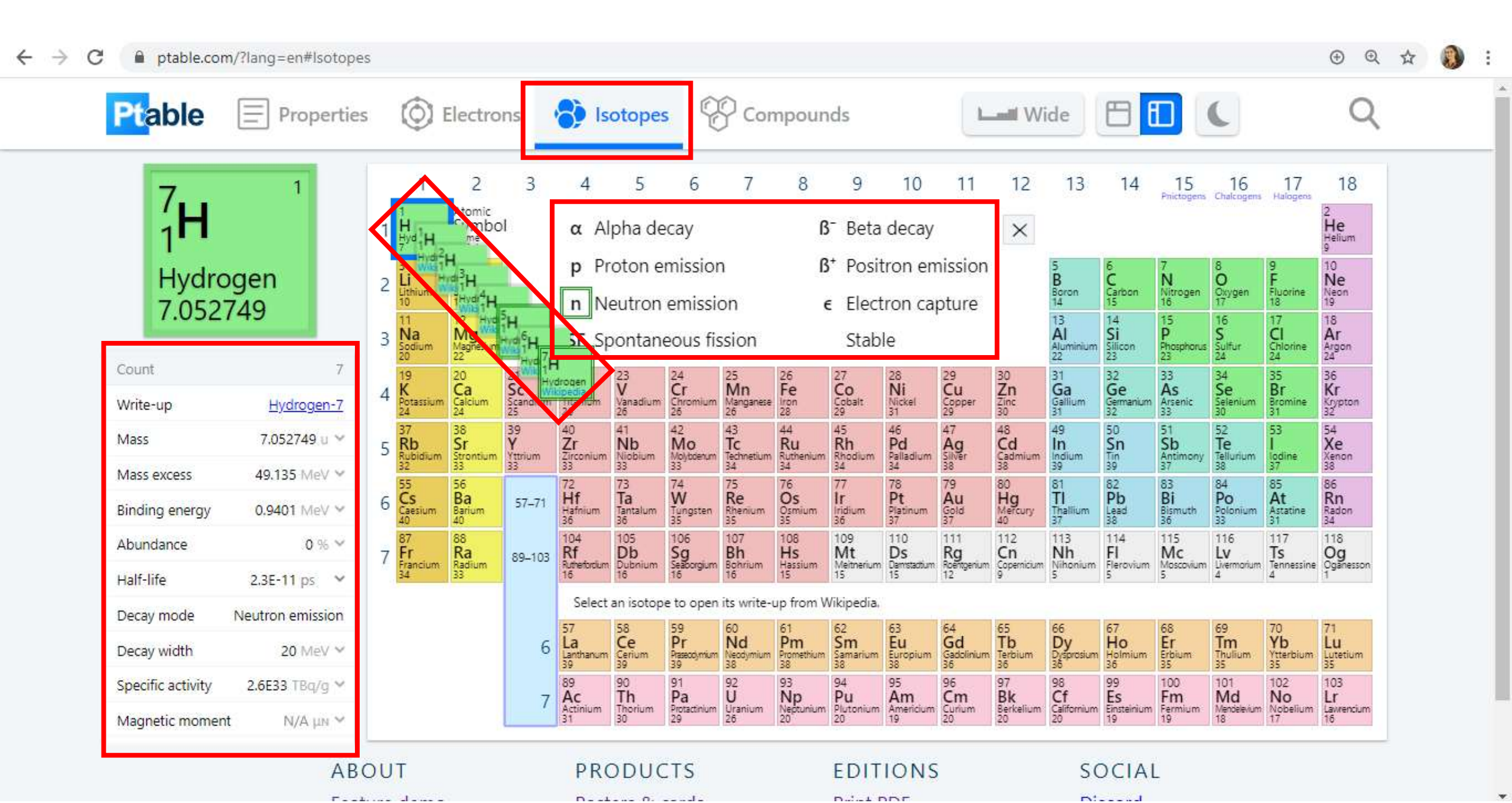
ELECTRONS - It displays the number of electrons in the orbitals and 3D structure of the elements



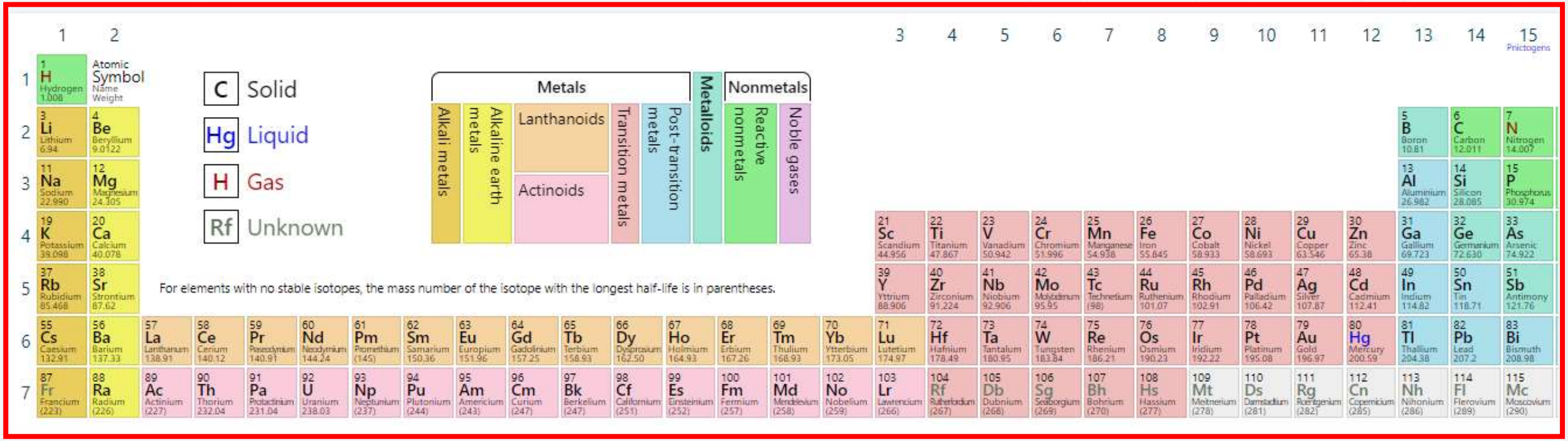
ISOTOPES - It displays the number of isotopes formed and various isotopic properties of the elements

COMPOUNDS - It displays the number of compounds formed and various matches for the selected elements

ptable.com/?lang=en#

Ptable | Properties | Electrons | Isotopes | Compounds | Wide

Temperature - + 0 °C 32 °F 273 K



ABOUT PRODUCTS EDITIONS SOCIAL

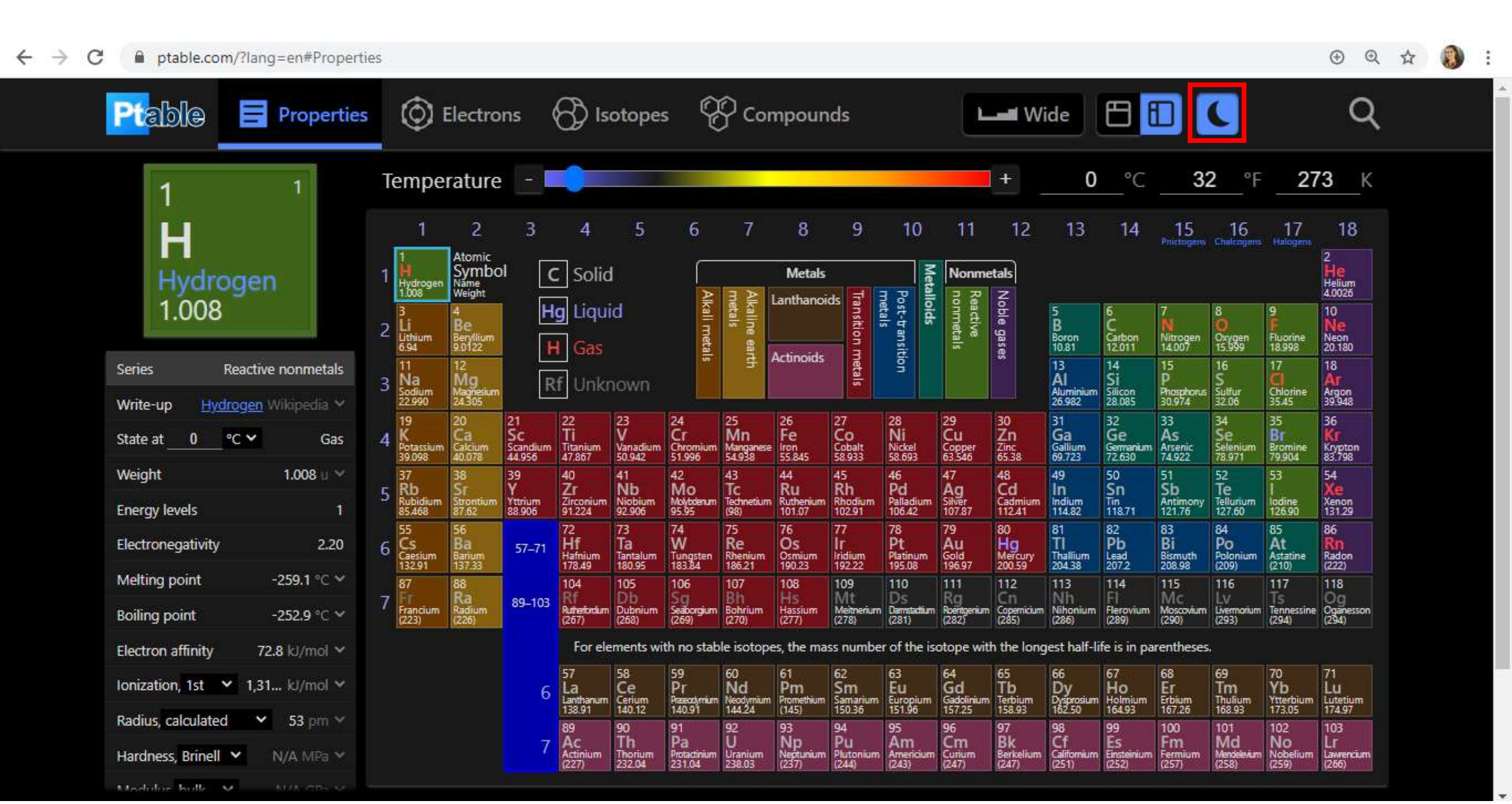
ptable.com/?lang=en#Properties

Ptable | Properties | Electrons | Isotopes | Compounds | Wide

1			1
H	Series: Reactive nonmetals	Melting point: -259.1 °C	Modulus, bulk: N/A GPa
Hydrogen	Write-up: Hydrogen Wikipedia	Boiling point: -252.9 °C	Density, STP: 0.0899 kg/m³
1.008	State at 0 °C: Gas	Electron affinity: 72.8 kJ/mol	Conductivity, thermal: 0.1805 W/mK
	Weight: 1.008 u	Ionization, 1st: 1,312.0 kJ/mol	Heat, specific: 14.300 J/kgK
	Energy levels: 1	Radius, calculated: 53 pm	Abundance, universe: 75 %
	Electronegativity: 2.20	Hardness, Brinell: N/A MPa	Discovered: 1766

Temperature - + 0 °C 32 °F 273 K

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15 Pnictogens	16 Chalcogens	17 Halogens	18
1	1 H Hydrogen 1.008	Atomic Symbol Name Weight	C Solid	Metals								Metalloids	Nonmetals					2 He Helium 4.0026
2	3 Li Lithium 6.94	4 Be Beryllium 9.0122	Hg Liquid	Alkali metals	Alkaline earth metals	Lanthanoids		Transition metals	Post-transition metals				Reactive nonmetals	Noble gases	5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180
3	11 Na Sodium 22.990	12 Mg Magnesium 24.305	H Gas			Actinoids									13 Al Aluminium 26.982	14 Si Silicon 28.085	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948
4	19 K Potassium 39.098	20 Ca Calcium 40.078	Rf Unknown	21 Sc Scandium 44.956	22 Ti Titanium 47.867	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.630	33 As Arsenic 74.922	34 Se Selenium 78.971	35 Br Bromine 79.904	36 Kr Krypton 83.798
5	37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.95	43 Tc Technetium (98)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 I Iodine 126.90	54 Xe Xenon 131.29
6	55 Cs Caesium 132.91	56 Ba Barium 137.33	57–71	72 Hf Hafnium 178.49	73 Ta Tantalum 180.95	74 W Tungsten 183.84	75 Re Rhenium 186.21	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.97	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)
7	87 Fr Francium (223)	88 Ra Radium (226)	89–103	104 Rf Rutherfordium (267)	105 Db Dubnium (268)	106 Sg Seaborgium (269)	107 Bh Bohrium (270)	108 Hs Hassium (277)	109 Mt Meitnerium (278)	110 Ds Darmstadtium (281)	111 Rg Roentgenium (282)	112 Cn Copernicium (285)	113 Nh Nihonium (286)	114 Fl Flerovium (289)	115 Mc Moscovium (290)	116 Lv Livermorium (293)	117 Ts Tennessine (294)	118 Og Oganesson (294)
				For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.														
6				57 La Lanthanum 138.91	58 Ce Cerium 140.12	59 Pr Praseodymium 140.91	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.96	64 Gd Gadolinium 157.25	65 Tb Terbium 158.93	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93	68 Er Erbium 167.26	69 Tm Thulium 168.93	70 Yb Ytterbium 173.05	71 Lu Lutetium 174.97
7				89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr


ptable.com/?lang=en#Properties
Ptable
Properties
Electrons
Isotopes
Compounds
Wide
1
1
H
Hydrogen
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Temperature
0 °C
32 °F
273 K
Series
Reactive nonmetals
Write-up
Hydrogen Wikipedia
State at 0 °C
Gas
Weight
1.008 u
Energy levels
1
Electronegativity
2.20
Melting point
-259.1 °C
Boiling point
-252.9 °C
Electron affinity
72.8 kJ/mol
Ionization, 1st
1,31... kJ/mol
Radius, calculated
53 pm
Hardness, Brinell
N/A MPa
Atomic
Symbol
Name
Weight
C Solid
Hg Liquid
H Gas
Rf Unknown
Metals
Alkali metals
Alkaline earth metals
Lanthanoids
Actinoids
Transition metals
Post-transition metals
Metalloids
Nonmetals
Reactive nonmetals
Noble gases
Pnictogens
Chalcogens
Halogens
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Ptable | Properties | Electrons | Isotopes | Compounds | Wide

Install app?

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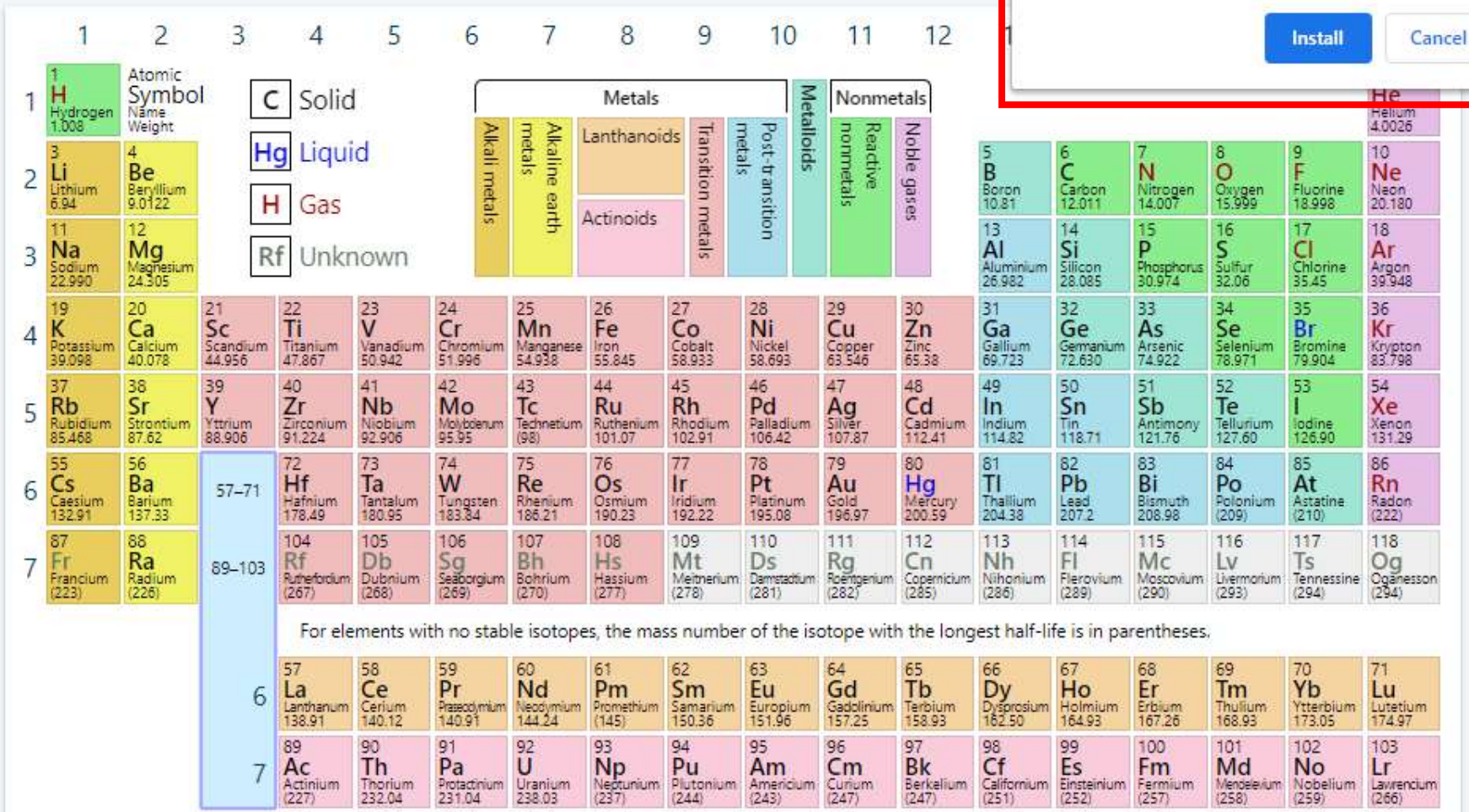
36
Kr
Krypton
83.798

2
8
18
8

Temperature - +

Series	Noble gases
Write-up	Krypton Wikipedia
State at 0 °C	Gas
Weight	83.798 u
Energy levels	2, 8, 18, 8
Electronegativity	3.0
Melting point	-157.36 °C
Boiling point	-153.22 °C
Electron affinity	0 kJ/mol
Ionization, 1st	1,35... kJ/mol
Radius, calculated	88 pm
Hardness, Brinell	N/A MPa
Modulus, bulk	N/A GPa



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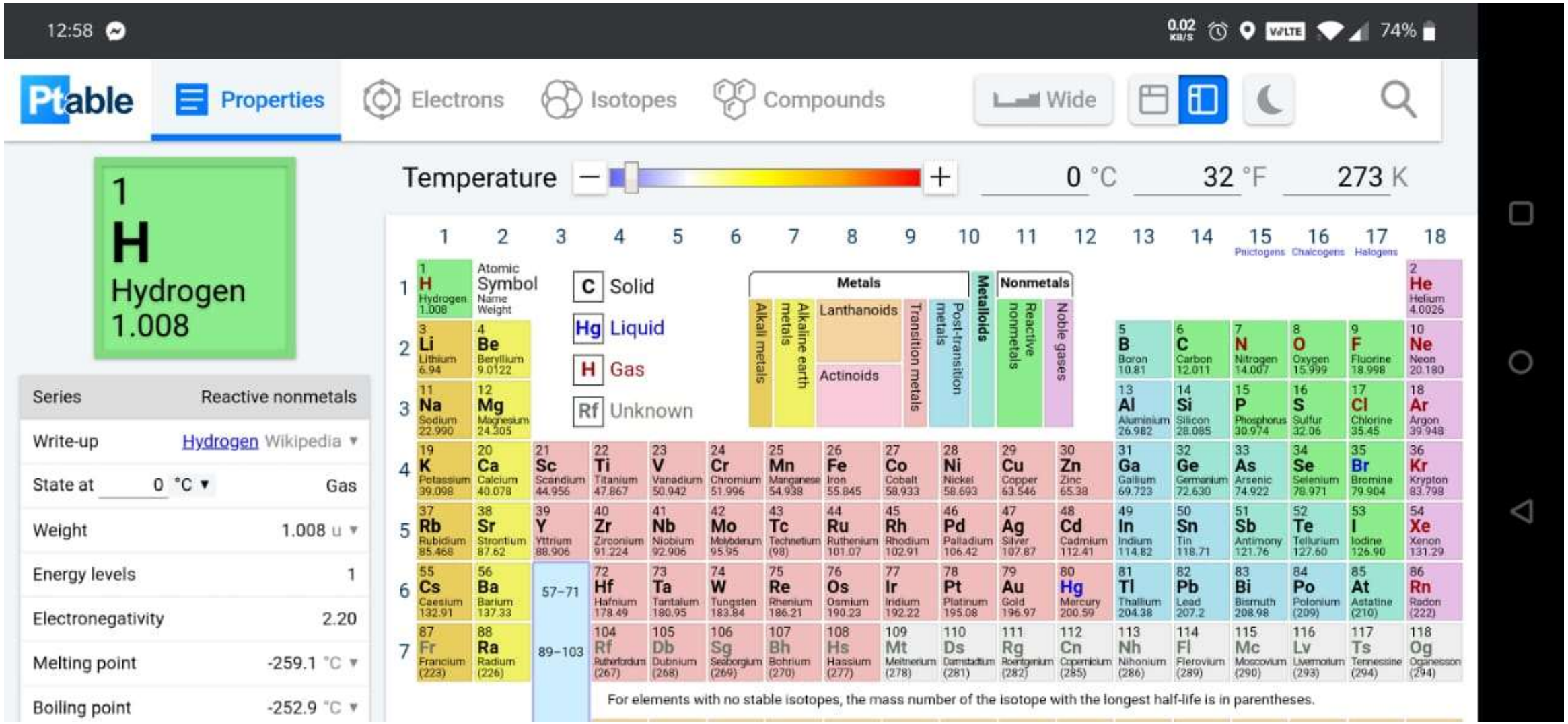

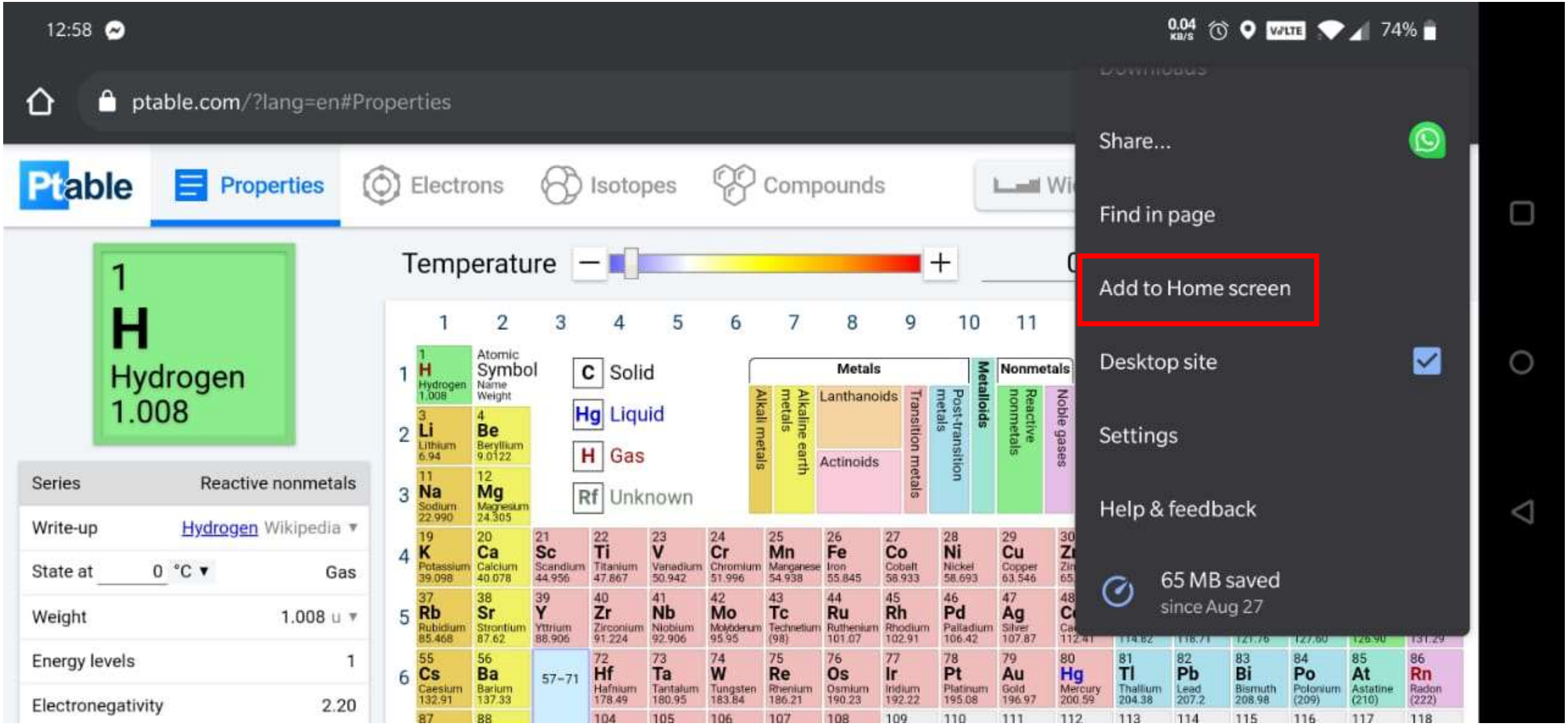
12:58
0.04 KB/S
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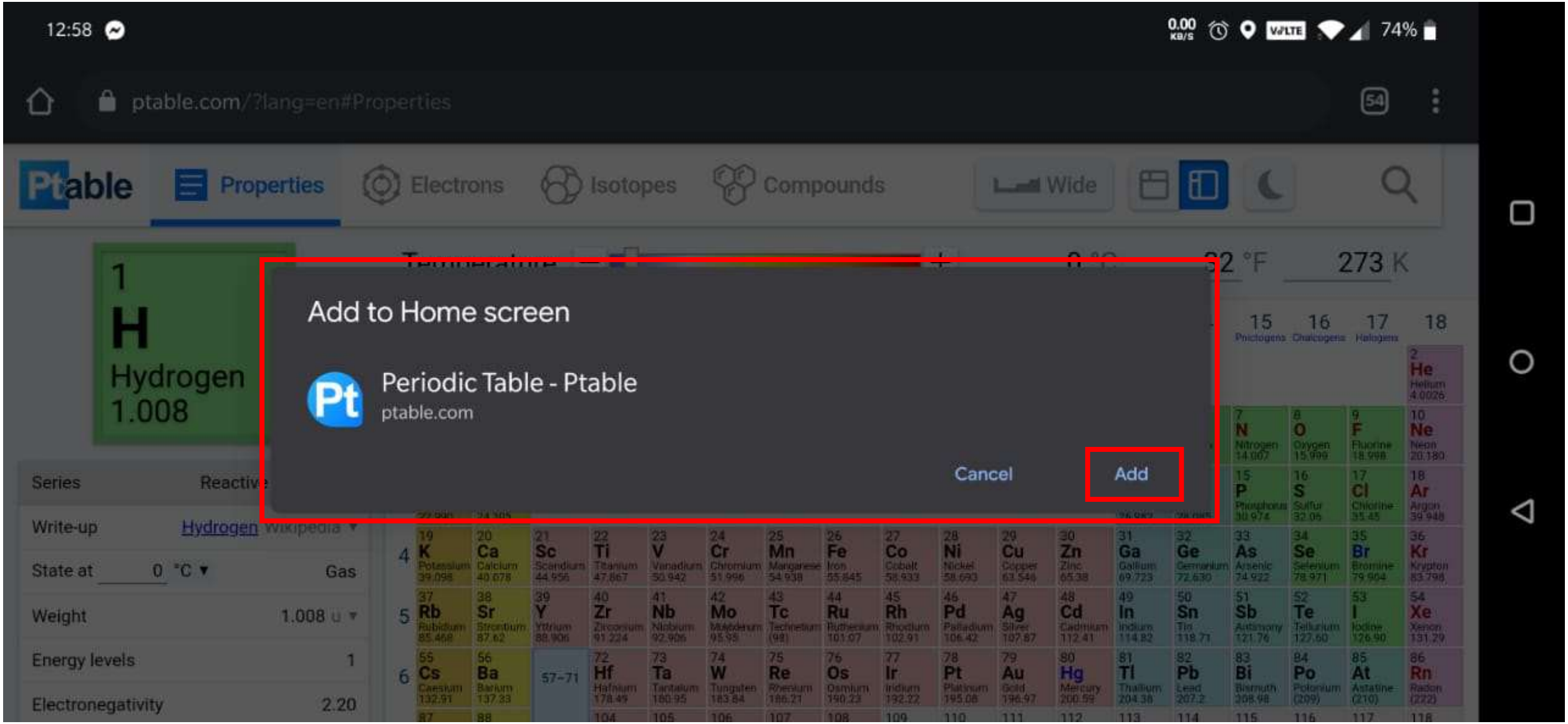
Hydrogen
1.008
Temperature
Series
Reactive nonmetals
Write-up
Hydrogen Wikipedia
State at 0 °C
Gas
Weight
1.008 u
Energy levels
1
Electronegativity
2.20
Atomic
Symbol
Name
Weight
C Solid
Hg Liquid
H Gas
Rf Unknown
Metals
Alkali metals
Alkaline earth metals
Lanthanoids
Actinoids
Transition metals
Post-transition metals
Metalloids
Nonmetals
Reactive nonmetals
Noble gases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Write-up
Hydrogen
State at
0 °C
Gas
Weight
1.008 u
Energy levels
1
Electronegativity
2.20

THANK YOU